

ization of many materials which could not otherwise be brought to the organisms which need them.

It has been calculated by Murray<sup>1</sup> that the total yearly run off of all the rivers of the earth is about 6500 cubic miles, carrying nearly 5,000,000,000 tons of dissolved mineral matter and prodigious quantities of sediment. The average composition of such water has been estimated as follows: —

	Parts per Million
Potassium as $K_2O$ . . . . .	2.40
Sodium as $Na_2O$ . . . . .	7.10
Lithium as $Li_2O$ . . . . .	0.20
Calcium as $CaO$ . . . . .	43.20
Magnesium as $MgO$ . . . . .	14.70
Manganese as $Mn_3O_4$ . . . . .	1.20
Iron as $FeO$ . . . . .	2.80
Aluminium as $Al_2O_3$ . . . . .	3.10
Silicon as $SiO_2$ . . . . .	16.40
Carbonic acid as $CO_2$ . . . . .	46
Phosphorus as $P_2O_5$ . . . . .	0.30
Nitric acid as $N_2O_5$ . . . . .	3.80
Sulphuric acid as $SO_3$ . . . . .	8
Chlorine as $Cl$ . . . . .	3.70
Ammonia as $NH_3$ . . . . .	0.07
Total mineral matter . . . . .	152.97

It is, of course, almost exclusively to these constant accessions that the ocean owes its salinity, which in the course of time has reached well-nigh inconceivable magnitude. The common salt alone in the oceans of all

<sup>1</sup> Russell, "Rivers of North America," p. 80.